Course Outline

Instructor
• Prof. Douglas Buchanan, P.Eng.
  E3–453 EITC
  (204) 474–9085
  Douglas.Buchanan@umanitoba.ca

Office Hours
• By appointment – please email.

Teaching Assistant
• Vaibhav Dubey
dubeyv@myumanitoba.ca
• Md. Iftekharul Islam
islamm@myumanitoba.ca

Contact Hours
• 4 credit hours
• Lectures:
  3 hours x 13 weeks = 39 hours
• Laboratories:
  3 hours x 5 weeks = 15 hours

Prerequisites:
• ECE 3670 Electronics 3
• PHYS 2152 Modern Physics for Engineers
• MATH 3132 Engineering Mathematical Analysis 3

Course Website:
http://umanitoba.ca/umlearn

ECE 3600 – Physical Electronics

Course Objectives
In this course we will cover basic solid-state theory; properties of semiconductors; principles of metal-semiconductor junctions, p-n junctions and transistors; and optoelectronic processes.

Course Content
The following topics will be covered:
• Electrons in a solid, Ohm’s law, Hall effect
• Electromagnetic fields in a solid, electron as a wave, de Broglie relation
• Quantum theory: Potential barriers and tunneling
• Free-electron model for metals, thermionic emission
• Band theory of solids, semiconductors
• Electrons, holes and effective mass
• Doping, law of mass action
• PN junctions: Rectification and capacitance
• Zener, avalanche and Schottky diodes
• MOSFETs, bipolar junction transistors (BJTs)
• Optoelectronics: Detectors and sources
• Dielectric materials: Polarization, ferroelectric and piezoelectric materials
• Magnetic materials: Properties, types, structures, uses.

Textbook

iClicker Reef app.
– The University has already purchased your subscription to the app for this course & term.

Other Resources
Any Foundation-year Physics text, eg. Halliday and Resnick, Serway, Tipler

Requirements/Regulations
• Attendance at lectures and laboratories is essential for successful completion of this course. Students must satisfy each evaluation component in the course to receive a final grade.
• It is the responsibility of each student to contact the instructor in a timely manner if he or she is uncertain about his or her standing in the course and about his or her potential for receiving a failing grade. Students should also familiarize themselves with the University’s General Academic Regulations, as well as Section 3 of the Faculty of Engineering Academic Regulations dealing with incomplete term work, deferred examinations, attendance and withdrawal.
• No programmable devices or systems (such as calculators, PDAs, iPods, iPads, cell phones, smart watches, wireless communication or data storage devices) are allowed in examinations unless approved by the course instructor.
• Students should be aware that they have access to an extensive range of resources and support organizations. These include Academic Resources, Counselling, Advocacy and Accessibility Offices as well as documentation of key University policies e.g. Academic Integrity, Respectful Behaviour, Examinations and related matters.

Important Dates
• Term Test
  Thursday, February 28th, 2019
  6:00PM–8:00PM
  E2–155 EITC
  E2–160 EITC
• Voluntary Withdrawal Deadline
  March 20th, 2019
• Spring Break
  February 18th–22nd, 2019
  No classes or examinations
Learning Outcomes

1. Summarize simple models that link the physical character of atoms and their bonding in solids to the bulk character of metallic semiconductor, dielectric and magnetic materials.

2. Identify how these basic descriptions of materials underpin simple operational descriptions of electronic devices and systems including, diodes, transistors, transducers and power transformers.

3. Demonstrate a mastery of these concepts through comparisons of how different materials/component designs can be used in examples including temperature sensing, optical detectors and data storage.

Expected Competency Levels

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Evaluation

Calculators are permitted for examinations. PDAs and other wireless electronic devices are forbidden. The supervisor may perform a hard reset on programmable calculators (programmable calculators are not required). Students must complete all the laboratories in order to be eligible to receive a passing grade.

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<th>Component</th>
<th>Value (%)</th>
<th>Method of Feedback</th>
<th>Learning Outcomes Evaluated</th>
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<td>Laboratories</td>
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<td>Final Examination</td>
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* Method of Feedback: F - Formative (written comments and/or oral discussion), S - summative (numerical grade)

Academic Integrity

Students are expected to conduct themselves in accordance with the highest ethical standards of the Profession of Engineering and evince academic integrity in all their pursuits and activities at the university. As such, in accordance with the General Academic Regulations on Academic Integrity, students are reminded that plagiarism or any other form of cheating in examinations, term tests, assignments, projects, or laboratory reports is subject to serious academic penalty (e.g. suspension or expulsion from the faculty or university). A student found guilty of contributing to cheating by another student is also subject to serious academic penalty.

Retention of Student Work

Students are advised that copies of their work submitted in completing course requirements (i.e. assignments, laboratory reports, project reports, test papers, examination papers, etc.) may be retained by the instructor and/or the department for the purpose of student assessment and grading, and to support the ongoing accreditation of each Engineering program. This material shall be handled in accordance with the University’s Intellectual Property Policy and the protection of privacy provisions of The Freedom of Information and Protection of Privacy Act (Manitoba). Students who do not wish to have their work retained must inform the Head of Department, in writing, at their earliest opportunity.

Note: These boundaries represent a guide for the instructor and class alike. Provided that no individual student is disadvantaged, the instructor may vary any of these boundaries to ensure consistency of grading from year-to-year.