



Course Outline

Instruction Team

- Dr. Fuji Jian (he/him),
P.Eng., Associate Professor
E1-352 EITC
(204) 474-7965
Fuji.Jian@umanitoba.ca

Student Hours

- Monday 2:30 – 4:20 PM
Individual assistance is
always available by
appointment – stop by!

Teaching Assistant

- Kenton McCorquodale-
Bauer
mccorquk@myumanitoba.ca
- Lecture location: E2-350
Lecture time: MWF 9:30-
10:20 am
- Tutorial Location: EITC E2
165
Tutorial time: M 2:30 -4:20
pm

Contact Hours

- 3 credit hours per week
Lectures:
3 hours x 12 weeks = 36
hours
Tutorial:
2 hours x 12 weeks = 24
hours

Course Website:

<http://umanitoba.ca/umlearn>

BIOE 2110 Transport Phenomena A01

Fall 2022 (10973)

Credit Hours: 3

Prerequisites and how they apply to this course

ENG 1460 Thermal Sciences (or equivalent course such as thermodynamics) provides students with an introduction to the thermodynamics (such as conservation of energy and efficiency of heat engines); the fundamentals of analyzing systems involving heat and energy; and application of mathematics and science to engineering problem solving. BIOE 2110 will build upon these experiences. Math 1500 (1510) and Math 1700 (or 1710) provide students with the knowledge of calculus. This knowledge will help students to understand the mathematical models (equations) of heat transfer.

Students' Learning Responsibilities

You are expected to learn the materials covered in lectures, tutorials, and assignments. Attendance for lectures and tutorials is strongly expected. Submission of assignments is to be done after tutorial or class time on the date mentioned in the assignments. To benefit the most from this class, you must be willing to review the course materials and participate in discussions about the material preferably during class or tutorial. While it is the instructor's responsibility to inform you changes in due dates, assignment material, etc..., it is your responsibility to read your university email regularly at least daily. Please also respect your fellow students during class, such as mute your cell phone.

Why this course is useful?

The problem-solving skills developed in this course are key to understanding fundamental principles of heat and mass transfer, numerical solving skill, model development and analysis, and estimation of complex systems through simpler models.

Who should take this course?

This is a required course in the Biosystems Engineering program and most engineering programs.

How this course fits into the curriculum?

It is intended that students take this course during or after the second year after being accepted into the Department of Biosystems Engineering. The general applied mathematical problem-solving skills are beneficial in most engineering science and design courses. The comprehension of heat transfer, steady state heat flow, conduction, convection, and radiation is key to understanding material properties and energy consumption in bioprocessing, alternative building systems, biomedical device components and agricultural systems.

- **Early Withdrawal Deadline**
September 20, 2022
- **National Day for Truth and Reconciliation**
September 30, 2022
No classes or examinations
- **Thanksgiving**
October 10, 2022
No classes or examinations
- **Fall Term Break**
November 7-10, 2022
No classes or examinations
- **Remembrance Day**
November 11, 2022
No classes or examinations
- **Voluntary Withdrawal Deadline**
November 22, 2022
- **Safety Test Deadline**
December 12, 2022
- **Last Day of Classes**
December 12, 2022
- **Midterm exam I:** Oct. 17
(Monday)
- **Midterm exam II:** Nov. 21
(Monday)
- **Final examination:**
Arranged by the
Department

Course Objectives

The intent of this course is to introduce students to:

1. Model and analyze heat and mass transfer problems.
2. Use numerical computation, estimation from graphs, and mathematical models (equations) to obtain information to solve problems.
3. Document technical computations to ensure that assumption, solution technique, and calculations can be validated.
4. Investigate the impact of materials' physical properties on heat and mass transfer (e.g., conductivity, thermal diffusivity, convection coefficient, and emissivity).

Undergraduate Calendar Description

Principles of heat transfer, solar radiation, psychometrics, molecular diffusion, mass transfer, and refrigeration and their application to biosystems.

Instructional Methods

Learning is most effective when both the teacher and the students are engaged in subject materials. The role of the instructor, therefore, is to create an environment that facilitates student engagement (and therefore learning). In this course, most dissemination of information will occur using the traditional lecture format (PowerPoint slides and writing on white board). A substantial portion of the content will be “presented” experientially through assignments and in-class examples. Therefore, you will be expected to prepare for class by reading the course text, completing the assignments, and questioning the professor whenever possible.

Course Content:

1. Introduction
2. Steady-state heat conduction
3. Transient heat transfer
4. Forced convective heat transfer
5. Free convection
6. Radiative heat transfer
7. Solar radiation
8. Psychometrics and mass diffusion
9. Heat exchangers

Texts, Readings, Materials

Textbook(s): Bergman T.L., Lavine A.S., Incropera F.P., and D.P. DeWitt, 2011. Fundamentals of heat and mass transfer, 8th ed., John Wiley & Sons, Inc., Toronto.

Supplementary Reading

Borgnakke, C., Sonntag, R.E., and G.J. Van Wylen 2009. Fundamentals of Thermodynamics, 7th. ed., John Wiley & Sons, New York.

Additional Materials: Supplied by instructor.

Grading Scale

Note: These boundaries represent a guide for the instructor and class alike. Provided that no individual student is disadvantaged, the instructor may vary any of these boundaries to ensure consistency of grading from year-to-year.

Letter	Mark
A+	90–100
A	85–89
B+	80–84
B	75–79
C+	65–74
C	60–64
F	< 50

Accreditation Units

- Mathematics: 0%
- Natural Science: 0%
- Complementary Studies: 0%
- Engineering Science: 100%
- Engineering Design: 0%

Description of Assignments

There will be 10 written assignments throughout the semester. Each assignment will consist of questions. They are to be answered in full using the methodology described in the class and assigned text. A question will be considered correct only if it is answered in the appropriate format, using appropriate significant digits and has an answer(s).

Description of Examinations

Two midterm examinations will be scheduled in the October and November, respectively. A final examination will be scheduled at the end of the semester. The examination will test the student's knowledge of the lecture material covered in this course.

Late Assignments

Will not be accepted and will receive a zero grade.

Missed Assignments

Will receive a zero grade unless student has a valid medical certificate or compassionate reason (see Missed Exams).

Missed Exams

There will be no "make-up" midterms. If a midterm examination is missed and the student has a valid medical certificate or compassionate reason (i.e., death of an immediate family member), the grade will be transferred to the final. Students who miss a midterm examination without a valid reason will receive a grade of zero for the midterm examination.

Learning Outcomes

At the end of the course, the student should be able to:

1. Model and solve heat transfer problems as electrical analogue circuitry,
2. Understand the correspondence between electrical and heat-based systems,
3. Apply steady state assumptions to determine the solutions,
4. Apply geometry simplification to generate and solve models of systems involving heat transfer,
5. Identify assumptions of typical heat transfer modelling,
6. Identify mechanisms of heat and mass transfer under a given situation,
7. Identify and compute thermophysical material properties, and
8. Compute estimations of heat loss from structures.

Grade Evaluation

The grade for this course will be based on assignments, two midterm examinations, and a final examination. The final grade is the combination of the following grades:

35% Final Examination (3hrs)

25% Midterm I (90 min)

25% Midterm II (90 min),

15% Assignments.

Expected Competency Levels

Graduate Attributes

- KB: A knowledge base for engineering
 PA: Problem analysis
 IN: Investigation
 DE: Design
 ET: Use of engineering tools
 IT: Individual and team work
 CS: Communication skills
 PR: Professionalism
 IE: Impact of engineering on society/environment
 EE: Ethics and equity
 EP: Economics and project management
 LL: Life-long learning

Com. Level	KB	PA	IN	DE	ET	IT	CS	PR	IE	EE	EP	LL
1												
2												
3	X											
4		X										
5		X										
6												

Graduate Attributes Assessed

- KB.3 – Application (Ability to apply knowledge in a new situation) of heat transfer knowledge to engineering problem.
 PA.4 – Analysis (Able to break problem into its components and establish relationships) heat transfer problem by using the knowledge delivered in the class.
 PA.5 – Synthesis (Able to combine separate elements into a whole) heat transfer problem in engineering application.

Competency Levels (Com. Level)

- 1 - Knowledge (Able to recall information)
- 2 - Comprehension (Ability to rephrase information)
- 3 - Application (Ability to apply knowledge in a new situation)
- 4 - Analysis (Able to break problem into its components and establish relationships.)
- 5 - Synthesis (Able to combine separate elements into a whole)
- 6 - Evaluation (Able to judge the worth of something)

Evaluation

Component	Value (%)	Assessor	Method of Feedback*	Learning Outcomes Evaluated	I/T*
Assignment	15	TA	F	1 to 8	I/T
Middle exam I	25	Dr. Jian	F, S	1 to 4	I
Middle exam II	25	Dr. Jian	F, S	4 to 8	I
Final exam	35	Dr. Jian	F, S	1 to 8	I

* Method of Feedback: **F** - Formative (written comments and/or oral discussion), **S** - summative (numerical grade)

** I/T: **I** – Individual effort, **T** – Team effort

Academic Integrity

Plagiarism or any other form of cheating in examinations, term tests or academic work is subject to serious academic penalty. Cheating in examinations or tests may take the form of copying from another student. Exam cheating can also include exam impersonation. A student found guilty of contributing to cheating in examinations or term assignments is also subject to serious academic penalty. Students should acquaint themselves with the University's policy on plagiarism, cheating, exam impersonation and duplicate submission.

Use of Third Party Detection and Submission Tools

Electronic detection tools may be used to screen assignments in cases of suspected plagiarism.

Group Work Policies:

All assignments are to be done as individuals and the University's policy plagiarism does apply to assignments (see Academic Integrity above).

Additional Policies: The tutorials are not optional.

Requirements/Regulations

- No programmable devices or systems (such as calculators, PDAs, iPods, iPads, cell phones, smart watches, wireless communication, or data storage devices) are allowed in examinations unless approved by the course instructor.
- All email communication must conform to the Communicating with Students university policy.

[Communicating with Students](#)

- Attending lectures and laboratories is essential for the successful completion of this course.
- Self-declaration forms may be completed for missed tests, exams, or assignments during short-term absences (≤ 72 hours) for extenuating circumstances. Students don't need to share personal information about their situation beyond declaring the nature of the extenuating circumstance on the self-declaration form.

[Self-Declaration Form for Brief or Temporary Absence](#)

- This form cannot be used for planned absences like vacations. It is also not to be used for longer-term absences, or ongoing circumstances (e.g., Authorized Withdrawals, Leaves of Absence, or other accommodations), which will still require additional documentation.

[Self-Declaration Policy for Brief or Temporary Absences](#)

- It is the responsibility of each student to contact the instructor in a timely manner if he or she is uncertain about his or her standing in the course and about his or her potential for receiving a failing grade. Students should familiarize themselves with the University's *General Academic Regulations*, as well as Section 3 of the Faculty of Engineering *Academic Regulations* dealing with incomplete term work, deferred examinations, attendance, and withdrawal.

[General Academic Regulations](#)

[Engineering Academic Regulations](#)

- Students should be aware that they have access to an extensive range of resources and support organizations. These include Academic Resources, Counselling, Advocacy and Accessibility Offices as well as documentation of key University policies e.g. Academic Integrity, Respectful Behaviour, Examinations and related matters.

[Supplemental Resources](#)

Retention of Student Work

Students are advised that copies of their work submitted in completing course requirements (i.e. assignments, laboratory reports, project reports, test papers, examination papers, etc.) may be retained by the instructor and/or the department for the purpose of student assessment and grading, and to support the ongoing accreditation of each Engineering program. This material shall be handled in accordance with the University's *Intellectual Property Policy* and the protection of privacy provisions of *The Freedom of Information and Protection of Privacy Act (Manitoba)*. Students who do not wish to have their work retained must inform the Head of Department, in writing, at their earliest opportunity.

Copyright Notice

All materials provided in this course are copyright and are provided under the fair dealing provision of the Canadian Copyright Act. This material may not be redistributed in any manner without the express written permission of the relevant copyright holder.

[Copyright Office](#)